Stark Effect

Motivation

Search for permanent electric dipole moments of atoms (Fr) Parity nonconservation (Cs forbiden transition)

-----> Tests of Standard Model

needed

Stark Effect Interaction between electric field and atoms

Spectroscopic data of polarizability Particularly for alkali atoms





Systematic study

Relativistic effect Core contribution

Precise polarizability data : needed For tests of calculation

Motivation



By high-resolution spectroscopy

Theory

Stark Effect

Theory

Stark shift < HFS

$$\gamma(J,F,m) = \frac{\left[3m_F^2 - F(F+1)\right] \left[3X(X-1) - 4F(F+1)J(J+1)\right]}{(2F+3)(2F+2)F(2F-1)J(2J-1)}$$

$$X = F(F+1) + J(J+1) - I(I+1)$$

- J: electronic angular moment
- F: total angular moment of atom
- I: nuclear spin
- **m_F: magnetic quantum number**



Experimental Setup



Separated Electrode System



Compact Electrode System



Internal Focus System



Interaction region~<1mm

Transitions Measured for Rb



HFS Spectrum of Rb D1 Transition



Stark Spectrum of Rb D1 transition



Dependence of Stark shift on electric field D1 transition



HFS Spectrum of Rb D2 Transition



Stark Spectrum of Rb D2 Transition



Relative Intensity [arb.units]

Dependence of Stark Shift on Electric Field D2 Transition



Scalar & Tensor Polarizability Rb D1, D2 Transition



a: Richard Marrus, Douglas Mccolm and Joseph Yellen : Phys. Rev. 147, 55 (1966).

b: Arthur Salop, Edword Pollack and Benjamin Bederson: Phys. Rev. 124, 1431 (1961).

Summary

- 1. High-resolution UV Laser Spectroscopy HFS Specific mass shift
 - **Electron density**
 - **Electronic configuration**
- 2. Stark Effect

Electrode system Rb D1 and D2 Stark shift and splitting Scalar and tensor polarizability